Vapor Intrusion Risk Pathway: Sampling & Analytical Issues

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This presentation is an excerpt of the vapor intrusion training that Dr. Hartman has been presenting to Federal & State regulatory agencies, DOD facilities, consulting groups, and stakeholders around the country. As of March 2013, this training has been given to over 30 State Regulatory agencies, EPA-OUST, ITRC, Brazil & Australia. Training has also been given to many PRPs such as the major oil companies, DOD, & numerous consulting groups.

Lecture notes are at the bottom of each slide so that if played out as a hard-copy, the presentation can be a useful reference document.

Exclusion Criteria: A PVI Pathway Game Changer

- Step 1: Can Site Be Screened Out?
 - Based upon concentration & depth to source
 - In CA if site is an active fueling station
- Step 2: Can "Screen-Out Data" be Collected?
- Step 3: Do PVI Assessment

VI Assessments Much Simpler & Less Expensive



So called "exclusion criteria", meaning criteria to eliminate sites from further vapor intrusion assessment based upon source concentration and distance between the source & receptor have fundamentallychanged the process for investigating the VI pathway at petroleum sites.

Table 3 Required Vertical Separation Distance Between Contamination And Building Foundation, Basement, Or Slab.

Media	Benzene	ТРН	Vertical Separation Distance (feet)*
Soil	≤10	≤250	6
(mg/kg)	>10 (LNAPL)	>250 (LNAPL)	15**
Groundwater	≤ 5,000	≤30,000	6
(ug/L)	>5,000 (LNAPL)	>30,000 (LNAPL)	15**

The thresholds for LNAPL indicated in this table are indirect evidence of the presence of LNAPL. These thresholds may vary depending on site-specific conditions (e.g., soil type, LNAPL source). Investigators may have different experiences with LNAPL indicators and may use them as appropriate. Direct indicators of LNAPL also apply; these include measurable accumulations of free product, oily sheens, and saturated bulk soil samples.

*Vertical separation distance represents the thickness of clean (TPH ≤ 100 mg/kg), biologically active soil between the source of PHC vapors (LNAPL, residual LNAPL, or dissolved PHCs) and the lowest (deepest) point of a receptor (building foundation, basement, or slab).

EPA-OUST's soon to be released PVI guidance contained exclusion criteria for soil & groundwater contamination and NAPL in the latest draft.

^{**} EPA recommends that sub-slab monitoring be used to evaluate the risk of vapor intrusion whenever LNAPL is present in any sample and the vertical separation distance is less than 15 feet. When LNAPL is

CA Low-Threat Closure Policy

Site Screens Out from VI Pathway:

- If 30' of Biozone, NAPL screens out
 - Vertically & horizontally
- If 10' of Biozone, benzene up to 1000 ug/L
- If 5' of Biozone, benzene up to 100 ug/L

Bioattenuation zone: TPH-soil < 100 mg/kg

Note: O₂ not Required



California released the low threat closure policy (LTCP) last summer (2012). It contains exclusion criteria similar to, but more conservative, than EPA's.

Allowable Benzene in GW 1e-6 risk, Residential Scenario

DTSC VI Guidance:
 0.084 ug/m3 / 0.002 = 0.42 ug/L / 0.2 = 0.21 ug/L

~5000x lower than LTCP of 1000 ug/L

New EPA OSWER Guidance:
 0.31 ug/m3/0.001 = 0.31 ug/L/0.2 = 1.55 ug/L

~3300 times lower than EPA-OUST Value of 5000 ug/L



The screening levels for dissolved benzene contamination in the CA-LTCP are 5000 times higher than CA-EPA levels in their VI guidance.

The screening levels for dissolved benzene contamination in the EPA-OUST draft guidance are 5000 times higher than in the EPA-OSWER VI guidance.

Clearly these new policies have major ramifications on the number of PVI sites requiring VI assessments.

Low-Threat Closure Policy The Power of Oxygen

CA-LTCP: If oxygen in soil gas >4%:

- Separation distance drops from 10' to 5' for benzene up to 1000 ug/L
- Soil gas SLs increase by 1000x!

TPH-soil required for all scenarios but O2 not



Oxygen plays a key role in some State policies. In CA, the presence of oxygen decreases the separation distance by a factor of 2 and increases soil vapor screening levels by 1000 times!

Soil Gas Allowed Levels

Benzene in Soil Gas, Residential Receptor, 1-6 Risk

	RBSL (ug/m ³)
DTSC Sub-slab	1.6
CHHSL	37
DTSC – Step 5	42
CA Low-Risk Policy: O2<4%	85
CA Low-Risk Policy: O2>4%	85,000

Ambient levels: 1 to 10 ug/m3



This table summarizes the various soil gas screening levels for benzene in CA. Note:

- 1) The sub-slab value is lower than ambient levels.
- 2) The presence of oxygen increases soil gas screening values by 1000 times.

NJDEP Gasoline Exclusion Criteria

VI Investigation is not required when:

- ≥10 ft between water table and foundation and benzene in GW is ≤1,000 µg/L; or
- ≥5 ft between seasonal high water table and foundation, oxygen levels measured at ≥2% (v/v), and benzene in shallow GW is ≤1,000 µg/L.

O₂ Drops Separation Distance from 10' to 5'

The presence of oxygen drops the separation distance in NJ by a factor of 2.

Step 2: "Screen-Out Data"

- Soil Phase TPH
- Presence of NAPL Go to ITRC PVI document
- Oxygen
 - CA & NJ for sure
- Soil Gas VOC Data not required at this step

Costs a Fraction of Typical Soil Gas Survey



Some sites may not have the soil phase data or oxygen data to screen out. Rather than perform a full VI investigation, an intermediate and less costly step is to collect the additional data to determine if the site can be screened out. These data will be primarily soil phase data and oxygen data. These data are far less expensive to obtain than typical vapor intrusion VOC data..

Step 3: PVI Specific Sampling Issues

- Soil Gas VOC Analysis
 - Benzene, ethylbenzene & naphthalene
 - TPH??
- Might Need to Sample <5' bgs
 - If samples >5' bgs exceed allowable levels
 - How to know? On-site analysis best
 - If not, collect samples anyway
- Always Collect Oxygen Data

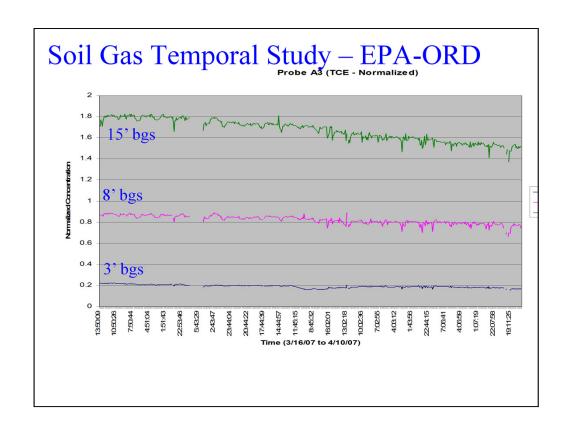


There are some differences in soil gas sampling for petroleum hydrocarbon VOCs than for chlorinated solvents.

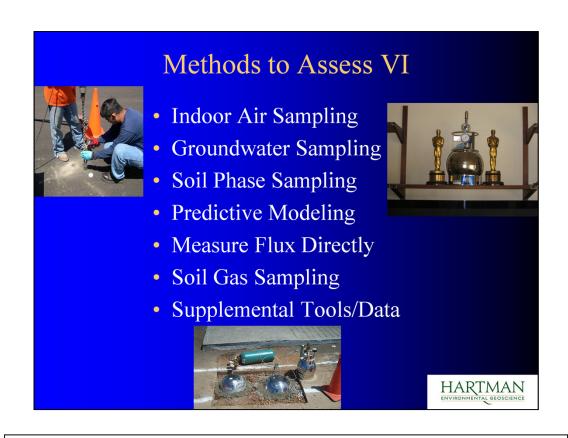
The COCs need to be determined and vary from State to State.

If samples at deeper depths exceed allowable values, shallower samples (<5' bgs) may need to be collected to document the effect of bioattenuation.

Oxygen data should always be collected to the document presence of the aerobic zone.



This is a plot of data recently collected for an EPA funded study by an automated instrument at at Vandenberg AFB site from three probes at the same location but at different depth (3', 8, & 17' bgs). This plot consists of over 500 points per probe collected once per hour over a 4 week period from mid March to mid April 2007. The soil gas concentrations varied by less than 10% over these four days even for probes only 3 feet below the surface.



These are the most common methods used in a vapor intrusion assessment.

The Most Important Ingredient

- Experience:
 - Consultant
 - Collector done soil gas before?
 - Lab certified for methods?
 - Regulator
 - Public
 - -YOU!



The most important ingredient for cost effective and efficient VI investigations is the experience of the people out in the field. Is the consulting firm experienced at VI investigations? Is it a routine part of their services or an occasional part? Do they put experienced staff in the field who can think or junior staff who aren't experienced?

The same applies to their subcontractors. Does the driller know how to collect soil gas samples? Is the lab experienced at analyzing vapors and does it hold an accreditation?

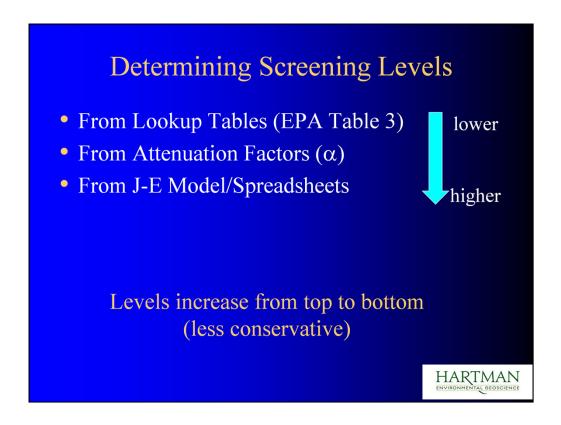
Most Common VI Bloopers

- Unit Confusion
 - Assuming ug/L equivalent to ppbv
 - Assuming ug/m3 equivalent to ppbv
- Screening Levels
 - Comparing to generic screening levels
 - Not calculating correct levels
- Sampling & Analysis Errors
 - Program design: soil gas? GW? SS? IA?
 - Using wrong hardware
 - Using wrong analysis



The 3 most common mistakes made by practitioners in the vapor intrusion arena:

- 1) Confusing units
- 2) Using incorrect screening levels
- 3) Sampling & analysis errors.



lowest (most conservative) levels. The J-E Model gives the highest (least conservative) levels.

Three methods are typically used to determine screening levels. The first method listed gives the

RBSLs From Attenuation Factors

For Soil Gas:

$$C_{sg} = C_{indoor}/\alpha_{sg}$$

For Groundwater:

$$C_{gw} = C_{indoor}/(H*\alpha_{gw})$$

Example: C_{indoor} benzene = 0.31 ug/m3 (1e-6 risk)

$$C_{sg}(5') = 0.31/0.002 = 155 \text{ ug/m}3$$

$$C_{gw} = 0.31/(0.20*\ 0.001) = 1550\ ug/m3 = 1.55\ ug/L$$



By using attenuation factors, one can calculate target levels for soil gas and groundwater starting from the acceptable indoor air concentration.

This is the method the EPA guidance allows to determine screening levels in the soil gas or groundwater.

RBSLs from Models

- Johnson-Ettinger Most Common
 - GW, soil, soil gas spreadsheets
 - Least conservative RBSLs
 - No bioattenuation component
- Biovapor
 - J-E model with bioattenuation added
 - In Beta testing by EPA
 - Now available from API



groundwater, soil gas, and even soil phase data. The Johnson-Ettinger model/spreadsheet is the most common. API is releasing a version that includes bioattenuation.

Several models are available that allow you to calculate screening values for

Allowable Soil Gas Levels (Benzene 1e-6 Risk, residential)

State	Alpha	1/Alpha	Risk Based Level (ug/m³)
EPA Now	0.002	500	155
EPA 2012?	0.1	10	3.1 (gulp!)
CA	0.002	500	42
NJ	0.05	20	16
MO			118,000
TN	0.0013	780	2,414
CT	0.1	10	192

A summary of the allowable benzene levels in soil gas shows large variation and illustrates the main points: the new EPA guidance is 50x more restrictive and allowable levels are variable from State to State.

Indoor Air Measurement

- Pros:
 - Actual Indoor Concentration
- Cons:
 - Where From?
 - –Inside sources (everything!)
 - Outside sources (exhaust)
 - –People activities NO CONTROL!
 - Time-intensive protocols
 - Snapshot, limited data points
 - Expensive!!



Measuring indoor air might seem to be the most direct and simplest approach, but it has its share of problems. The biggest problem is background sources of contaminants. Many commonly used household products contain some of the target compounds of concern. For example, benzene from consumer products, PCE from dry cleaned clothes, TCE from degreasing cleaners. In addition, the protocols are laborious, intrusive, offer little control, and are expensive. For these reasons, the EPA and many States shy away from this method, especially for PVOCs. However, this method may still be the method of choice if the contaminant of concern is not one commonly found in household products (e.g., 1,1 DCE).

	DRAFT: Vol:	atile Organ	nic Cor	nnounds	by EPA	TO-15			
		&P Mobile		•					
Analyte	Result	Reporting Limit	Units	Dilution Factor	Batch	Prepared	Analyzed	Method	Notes
DRAFT: Shaving Cream (E103030-01) Vapor	Sampled: 03-M	ar-11 Receiv	ed: 04-M	ar-11					
Carbon disulfide	136	31.5		~"~		4 4 0	",		
trans-1,2-Dichloroethene	ND	40.2) =	14()	ug/m	3	
Methyl tertiary-butyl ether (MTBE)	ND	18.3	"		-	1.0	45/111		
Vinyl acetate	ND	17.8			-				
1,1-Dichloroethane	ND	20.5			-	"			
2-Butanone (MEK)	ND	149	"		-	"			
n-Hexane	2590	17.8	**						
cis-1.2-Dichloroethene	ND	20.1							
Diisopropyl ether (DIPE)	ND	21.2	**						
Ethyl acetate	ND	91.2	**				1	Mobile	
Chloroform	ND	24.8	**				5 G e	ochemistr	,
2,2-Dichloropropane	ND	23.4					6	Inc.	
	ND	149					0	IIIC,	
Tetrahydrofuran	ND	21.2							
Ethyl tert-butyl ether (ETBE)	ND	27.6			"		20	/ 0	
1,1,1-Trichloroethane	ND	20.5		⊀en7	ene	= -3	39 ug	1/m3	
1,2-Dichloroethane (EDC)	ND	23.0		JUIL	CIIC	, ,	ع مير	, 1115	
1,1-Dichloropropene	389	16.2				-			
Benzene		31.9					**		
Carbon tetrachloride	ND 469	87.1							
Cyclohexane	37100	27.8	ug/m3	5	EC10305	04-Mar-11	04-Mar-11	EPA TO-15	
p-Isopropyltoluene			ug/m3		EC10305	04-;viar-11	04-ivini-11	EFA 10-13	
1,2-Dichlorobenzene	ND	61.0							
n-Butylbenzene	3000	27.8					0 0 0		
1,2-Dibromo-3-chloropropane	ND	49.0		- T)H=	-680	0001	ug/m3	
Naphthalene	104	26.6		II	II	UUU,		ug/1113	
1,2,4-Trichlorobenzene	160	37.6	-		-		-		
1,2,3-Trichlorobenzene	134	37.6						_	
Hexachlorobutadiene	89.2	54.1							
Xylenes (total)	ND	22.0	**			-	"		

An analysis of Barbasol shaving cream: Benzene & TPH at levels 1000x above indoor air screening levels!



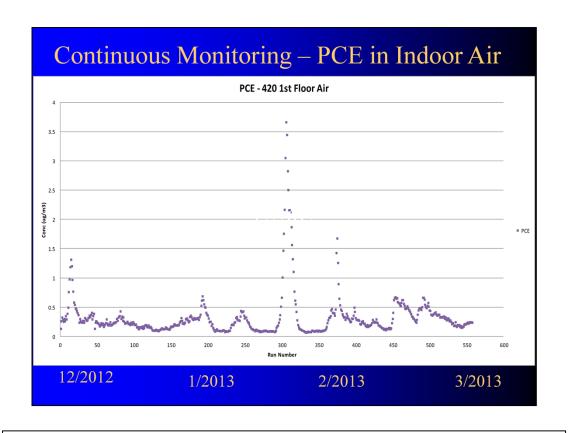
The hidden source of benzene & TPH contamination at many structures.

Indoor Air Sampling Lessons

- Always Collect Ambient Air Sample
- Hardware Issues
 - Blanks
 - Performance Fill at Proper Rate?
 - Fittings Tight? Cross-threaded?
 - Gauges on cans, not on flow chokes
 - Half filled = 2x rise in RLs

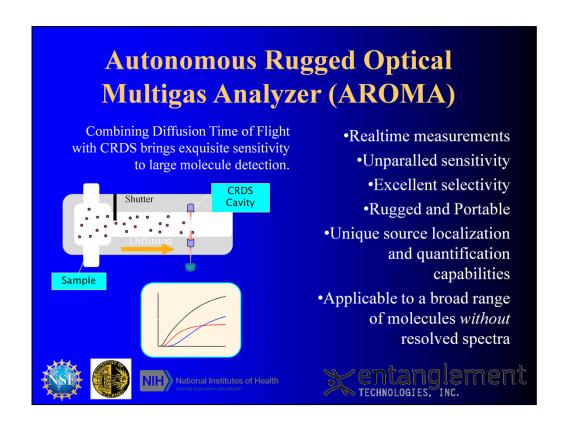


There are issues that need to be considered when sampling indoor air and when interpreting the data. Sampling issues include the hardware, time period for collection, and things as simple as the type of marker used to label the samples.

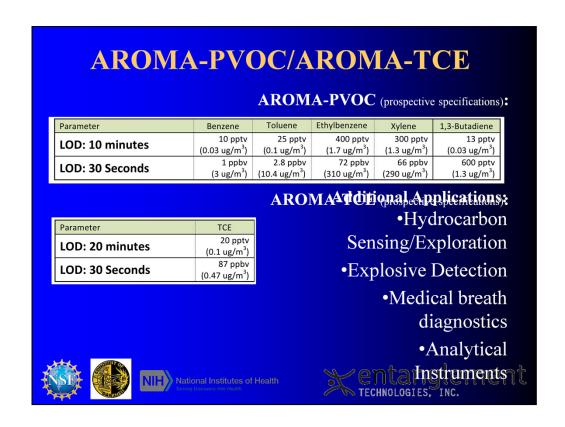


Continuous monitoring data of PCE in indoor air from a house. Sensitive chlorine detectors exist to allow this type of data to be collected.

Wouldn't it be nice to be able to do the same for benzene?



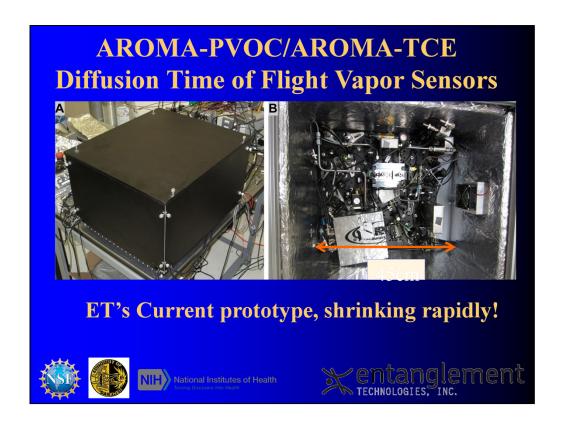
A new instrument being developed by a firm in CA may allow continuous monitoring for selected VOCs.



Note the listed specs for benzene and TCE.

Benzene at 0.03 ug/m3 is below 1e-6 risk level.

TCE at 0.47 ug/m3 in 30 seconds is at the 1e-6 risk level.



Currently the size of a microwave, but expected to shrink.

Which Soil Gas Method?

- Active?
- Passive? (limited use)
- Flux Chambers? (limited use)

Active method most often employed for VI

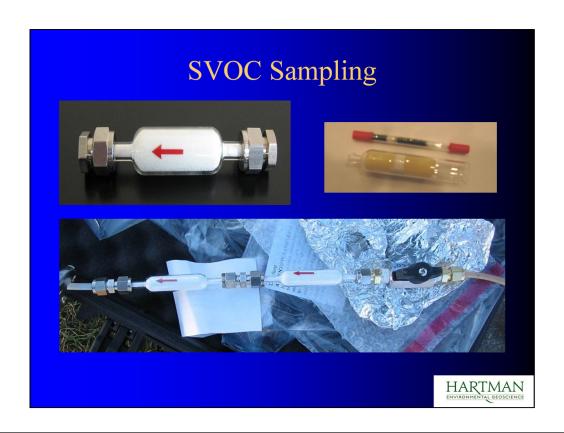


There are three types of soil gas methods. Active refers to actively withdrawing vapor out of the ground. It gives quantitative values. Passive refers to burying an adsorbent in the ground and letting the vapors passively contact and adsorb onto the collector. It does not give quantitative data and hence can not be used for risk applications, except for screening. Surface flux chambers were discussed previously.

The active method is the one most applicable to risk assessments.



A mobile trailer park with shallow (2' bgs) ground water. The prefec tsite for flux chambers.



A typical sampling arrangement used for collection of samples on adsorbants. Note the abundance of fittings and the need for duplicate cartridges for breakthrough. A very complicated set-up, prone to leaks.



New style adsorbent tubes are much smaller than bulky canisters and can't be broken.



Top photo: A typical sampling arrangement used for collection of samples on adsorbants. Note the abundance of fittings and the need for duplicate cartridges for breakthrough. A very complicated set-up, prone to leaks.

Bottom photo: A much simpler sampling arrangement for adsorbent tubes with better control on actual vapor volumes passed through the adsorbent.

TPH Compounds

- Recommended
 - BTEX (BE only drivers)
 - Methane
 - 1-2 dichloroethane (EDC) & 1-2 dibromoethane (EDB)
 - Naphthalene, 2-methyl naphthalene
- Some States:
 - Aliphatics (C5-C8 & C9-C12)
 - Aromatics (C9-C10)
 - TPHg different from the aliphatics



Petroleum compounds of concern vary from State to State. Consult the oversight agencies specifications.

Other Analytical Issues

- 1,3 Butadiene
 - False positive caused by i-butylene
 - Must have lab manually read ion chromatogram
 - Not on most agency soil gas target lists
- Naphthalene
 - − 8260, TO-15, TO-17 all used, but ...

TO-17 gets PVOCs, TPHg, TPHd in same run!!



Watch out for false detections of 1,3 butadiene.

Also, naphthalene is best analyzed by TO-17.

Parting Thoughts

Upcoming Conferences

- AWMA Annual Conf Chicago, 6/25/2013
- Cleanup 2013 Melbourne, Oz 9/16/2013

Looking for a Great PVI Guidance Template?

Look at the Australian PVI Guidance

HARTMAN ENVIRONMENTAL GEOSCIENCE

Upcoming vapor intrusion traini	ing.	